



# Edible Coating and Pulsed Light to Increase the Shelf Life of Food Products

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## Abstract

The application of edible coatings (EC) in combination with pulsed light (PL) treatments represents an emerging approach for extending the shelf life of highly perishable but high value-added products, such as fresh-cut fruits and vegetables. The surface of these products would benefit from the protective effects of ECs and the PL decontamination capability. This review describes in detail the fundamentals of both EC and PL, focusing on the food engineering principles in the formulation and application of EC and the delivery of efficient PL treatments and the technological aspects related to the food characterization following these treatments and discussing the implementation of the two technologies, individually or in combination. The advantages of the combination of EC and PL are extensively discussed emphasizing the potential benefits that may be derived from their combination when preserving perishable foods. The downsides of combining EC and PL are also presented, with specific reference to the potential EC degradation when exposed to PL treatments and the screening effect of PL transmittance through the coating layer. Finally, the potential applications of the combined treatments to food products are highlighted, comparatively presenting the treatment conditions and the product shelf-life improvement.

**Keywords** Edible coatings · Pulsed light · Combined treatments · Food preservation · Shelf-life extension · Nonthermal technologies

## Introduction

Mild preservation techniques have, nowadays, gathered a key role in many food productions. They are frequently used in replacement of heat treatments, to preserve the food nutritional and sensory properties, while ensuring microbial safety and a prolonged product shelf life [1]. In the case of products where heat treatments are not suitable, such as fresh or fresh-cut fruits and vegetables, or raw or cured meat and fish products, mild preservation techniques are chosen to avoid the use of synthetic preservatives, such as sulfites, benzoic acid, or its derivative salt, which are increasingly rejected by

consumers [2], and often represent the only option for wide-range distribution if the use of chemical preservatives is not allowed by food authorities [3, 4]. They can also be applied to contribute to improving the shelf life for enhanced distribution of perishable products or to compensate for logistics characterized by inefficient structures or inadequate cold chain in rural areas.

One possible, relatively simple approach, is represented by edible coatings (EC) [2]. Differently from edible films, which are first formed and then applied as a wrapping on the food product, ECs are applied in liquid form directly on the surface of the food to be coated, to exert a protective action against mechanical damage and chemical reaction, and to act as a moisture barrier. They also represent a physical barrier to microbiological attack. ECs have been traditionally used in food conservation, such as in the case of wax coatings for fruit, used since the twelfth century, chocolate coatings for confectionery, or lipid films for meat products [5]. In their modern version, ECs are applied as a very thin layer to minimally affect product appearance, to protect food products from deterioration processes, including oxidation, moisture absorption/desorption, chemical reactions, and microbial growth, as well as to improve their physical strength, reduce particle

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clustering, and possibly improve visual and tactile properties of food product surfaces [5]. These functions are mainly based on their capability to act as barriers against water or oil permeation and gas or vapor transmission [6]. Moreover, ECs can also be loaded with active ingredients, to integrate or develop properties that can aid in extending the shelf life, such as antimicrobial, anti-browning, antioxidant, coloring, and flavoring agents and even nutrients [2]. In particular, active coatings with antimicrobial properties are especially appealing, because they can be formulated with natural ingredients, and, in addition to the above-mentioned physical barrier properties, ECs also enable controlled release of natural antimicrobials [7–10].

The use of ECs in combination with other preservation technologies presents significant potential benefits to food preservation, within the hurdle technology approach, to attain mild but reliable preservation effects [3, 11]. The ECs have been tested in combination with different nonthermal technologies, such as high hydrostatic pressure, pulsed light [8],  $\gamma$ -radiation, ozone, UV light [9], and modified atmosphere packaging [12]. In some cases, synergistic effects (e.g., in combination with  $\gamma$ -radiation and modified atmosphere packaging [12] or high hydrostatic pressure [8]), in other cases, a simply additive effect (e.g., in combination with UV light [9]) or antagonistic effects (e.g., in combination with ozone [9] and pulsed light [8]) were observed on microbial stabilization. Pulsed light (PL) represents an extremely promising technology for the surface decontamination of food products [13], because of the microbial inactivation effect achieved by PL through structural damages caused by photophysical and photothermal effects, as well as the damage of the microorganism DNA, among others [14]. Therefore, PL has a recognized potential to be widely adopted in the food industry, as recently reviewed [13]. It is specifically suitable to be applied in combination with the application of ECs, to further contribute to extending the shelf life of high value-added food products, either by avoiding contamination after EC deposition, also through *in-package* treatments, or as a strategy to attain a fast initial reduction of the microbial load before the coating process.

This review describes the combined use of edible coatings and pulsed light treatments. It illustrates the key issues in the formulation of coating solutions and their deposition on food products, the addition of active ingredients to expand their functionalities, and the characterization methods, also as a function of the existing food applications. It also addresses the main technical aspects related to PL treatments, especially as a function of the target preservation process. Finally, it presents a critical analysis of the combination of ECs and PL treatments, discussing the possible process configurations and the interferences between the two technologies and surveying the existing literature on the topic.

## Edible Coatings

### Materials

#### Coating Materials

ECs can be fabricated with different materials, which can be classified into three categories: hydrocolloids, including polysaccharides and proteins; lipids; and composite materials, consisting of a combination of different hydrocolloids or hydrocolloids and lipids, to exploit the complementary functional properties of the different constitutive materials or overcome the respective drawbacks [5, 15, 16]. Table 1 reports a survey of the main film-forming materials, with the indication of the solvents needed for their application, and the main advantages and disadvantages of the ECs obtained with such materials. The selection of the coating materials and their additives is generally based not only on the desired technological properties, but also on market availability, cost-effectiveness, effects on the sensory attributes of the final product, and consumer acceptance [77, 78]. It is required that for market applications, the coating materials are generally regarded as safe (GRAS) and approved by regulatory agencies, such as the US Food and Drug Administration (US FDA), through the inclusion in the Code of Federal Regulations (CFR) - Title 21, or the European Food Safety Authority (EFSA), through the inclusion in the Food Additives database of the European Commission [79].

Notably, polysaccharide-based coatings include many water-soluble polymers, which are easy to prepare and deposit on food products, and usually are cost-effective [80, 81]. Generally, very hydrophilic polysaccharide coatings do not provide good water vapor barrier properties [82], but they exhibit selective permeability for O<sub>2</sub> and CO<sub>2</sub> and resist lipid migration [81, 83]. Among polysaccharide-based materials, alginates have been widely used in different products because of their properties and regulatory status [25–31]. Chitosan has also been intensively investigated because of its intrinsic antimicrobial activity [84] and because its properties can be further improved by chemical modification [9] or blending with other polymers. Many native and modified starches from plant sources have been used in the formulation of ECs because they exhibit very good encapsulation, film-making, and emulsification properties. However, their application as ECs is limited by their poor mechanical strength and barrier properties [85]. Most protein-based ECs show good hydrophilic but poor moisture barrier properties.

The application of lipids as ECs has not been favored, because they often exhibit microscopic pores, high O<sub>2</sub> solubility, and diffusivity, and cause undesirable organoleptic properties [86]. The combination of hydrocolloids and lipids could improve the structural integrity and characteristic functionality of the EC, as polysaccharide and proteins exhibit

**Table 1** Edible coating materials, solvent and coating solution preparation conditions, key properties and their advantages and disadvantages, product application examples, cost information, and regulatory status

Edible coating materials	Solvent for the coating solution	Advantages	Disadvantages	References
<b>1. Hydrocolloids</b>				
<b>1.1. Polysaccharides</b>				
Agar (E406) CAS No 9002-18-0	Water	Homogeneous, compact, and resistant layer	High water vapor and oxygen permeability, off-flavor production	[17–19]
Starches				
Cassava, manioc, potato, cornstarch	Water (high temperature)	Good preservation of quality attributes; protection against microbial contamination	Reduction in the visual quality and overall acceptability	[20–24]
Alginate (E400-E404) CAS No 9005-38-3	Water (high temperature)	Reduction of the browning process; protection against microbial contamination	Need for the addition of glycerol and cross-linking agent to improve mechanical properties; poor water vapor barrier	[25–31]
Cellulose derivatives				
Carboxymethyl cellulose (E466) CAS No 9000-11-7	Ethanol aqueous solution	Preservation of quality attributes	Reduction in bioactive content	[32–36]
Hydroxypropyl methylcellulose (E464) CAS No 9004-65-3	Ethanol aqueous solution	Transparent, sticky, and nonsmelly polymer; good gas barrier properties	Moisture susceptibility and high water vapor permeability; plasticizers needed to improve elasticity	[37, 38]
Chitosan CAS No 9012-76-4	Acetic or lactic acid aqueous solutions	Intrinsic antimicrobial activity, preservation of qualitative properties	Need for the addition of glycerol to improve the elasticity of film; impact on organoleptic properties; high cost	[25, 30, 34, 35, 39–41]
Carrageenan (E407) CAS No 9000-07-1	Water	Effective in reducing lipid oxidation; protection against microbial contamination	Poor film-forming properties; not effective in pre-serving quality attributes; high cost	[42, 43]
Gums				
Guar gum (E412) CAS No 9000-30-0	Water	Preservation of quality attributes; effective barrier for gases	Not effective in reducing microbial growth; need for the addition of glycerol to improve thermomechanical properties	[33, 44–46]
Gum arabic (E414) CAS No 9000-01-5	Water	Preservation of quality attributes; protection against microbial contamination	Need for the addition of glycerol to improve mechanical properties	[33, 47–50]
Xanthan gum (E415) CAS No 11138-66-2	Water	Preservation of quality attributes	Low water barrier properties; mechanical properties function of gum concentration	[33, 51, 52]
Gellan gum (E418) CAS No 71010-52-1	Water (high temperature)	Preservation of quality attributes; protection against microbial contamination	Need for the addition of glycerol and cross-linking agents to reduce film brittleness	[30, 53, 54]
Pectin (E440) CAS No 9000-69-5	Water	Preservation of quality attributes; high water vapor barrier properties	Lack of antibacterial activity	[55–58]
<b>1.2. Proteins</b>				
Gelatin CAS No 9000-70-8	Water (high temperature)	Good mechanical properties; transparent; excellent oxygen barrier and antioxidant properties; preservation of quality attributes	Low moisture barrier; need for the addition of cross-linking agent to strengthen the formed layer	[59, 60]
Casein CAS No 9000-71-9	Water (high temperature)	Good mechanical strength; low permeability to oxygen and nonpolar molecules; protection against microbial contamination	Need for use in combination with other materials to protect from oxidation	[61, 62]
Whey protein CAS No 92129-90-3	Water	Preservation of quality attributes; good oxygen barrier property; transparent	Need for the addition of glycerol to improve film flexibility; low moisture barrier	[26, 63, 64]

**Table 1** (continued)

Edible coating materials	Solvent for the coating solution	Advantages	Disadvantages	References
Zein	Ethanol	Preservation of quality attributes	Lack of antioxidant and antimicrobial activity	[65–67]
CAS No 9010-66-6				
Egg albumin	Water	Preservation of quality attributes; good water and fat barrier; bright and transparent film	Interaction with food product; high cost	[68]
CAS No 9006-59-1				
Soy protein	Water	Smooth and homogeneous coating layer; hydrophilic surface	Need for the addition of glycerol to improve mechanical properties; yellowish film coloration; high water vapor permeability	[69, 70]
CAS No 9010-10-0				
<b>2. Lipids</b>				
Beeswax (E901)	Water (high temperature)	Preservation of quality attributes; protection against microbial contamination	Effect on quality and sensory properties	[32]
CAS No 8012-89-3				
Caruba waxes (E903)	Emulsification in water	Preservation of quality attributes; hydrophobic coating; good moisture barrier	Sensory implications on food (off-flavors)	[71]
CAS No 008-015-869				
Coconut oil	Water (high temperature)	Preservation of quality attributes	Sensory implications on food (off-flavors)	[72]
Beeswax				
<b>3. Composites</b>				
Hydroxypropyl methylcellulose, beeswax, glycerol, GRAS salts	Emulsification in water	Synergic effect of hydrophobic and hydrophilic components; reduction in weight loss; good gas barrier properties	Interaction with the product; production of off-flavors; high cost	[73]
Chitosan, guar gum, glycerol	Acetic acid solution; emulsification in water	Preservation of quality attributes	High cost	[74]
Hydroxypropyl methylcellulose, beeswax, nanoclay, ginger EO	Emulsification in water (high temperature)	Preservation of quality attributes	Limited effect on product firmness; high cost	[75]
Guar gum, candelilla wax, glycerol, gallic acid	Emulsification in water (high temperature)	Preservation of quality attributes; good water vapor barrier	No significant effect on the accumulation of sugars; high cost	[76]

poor moisture barrier properties, whereas lipids exhibit poor gas barrier properties.

### Cross-linking Agents

Cross-linking by covalent and noncovalent bonds [87] of the coating polymers chains, particularly proteins and polysaccharides, can be achieved by the application of a cross-linking agent after depositing the polymer coating solution on the food surface, to form a more compact and resistant layer [88, 89] and to improve mechanical strength, chemical resistance, and thermal stability [90]. The most common cross-linking agents are symmetrical bifunctional compounds with reactive groups with specificity for functional groups present on the matrix polymer [91]. The most common cross-linking agents include glutaraldehyde for gelatin, cellulose derivatives, and chitosan [92–95];  $\text{Ca}^{2+}$  ions for alginate, pectin, and whey proteins [96–99]; citric acid for starch and cellulose derivatives [100–103]; and tannic acid for gelatin and chitosan [104, 105].

### Plasticizers

Plasticizers are additives rich in hydroxyl groups that contribute to preventing cracking during handling and storage, through the plasticization of the polymer network. The primary objectives of plasticizer addition include increasing the free volume or molecular mobility of biopolymers, decreasing intermolecular attractions between adjacent polymeric chains by reducing hydrogen bonding between polymers chains, bestowing flexibility, reducing brittleness, improving tear impact resistance, and regulating the flow of the coating material [106–111]. The most commonly used plasticizers are polyols (propylene glycol, glycerol, sorbitol, polyethylene glycol), oligosaccharides (glucose, fructose, and sucrose), and water (known as an excellent plasticizer [112]). Glycerol is particularly suitable and often used in hydrocolloid-based coatings [113, 114].

### Surfactants

Adhesion of hydrophilic ECs on hydrophobic food surface is inherently poor because the different chemical nature of the two surfaces makes it difficult to obtain a uniform EC [16, 115]. Surface active agents, such as surfactants, are frequently added into coating formulations to improve surface adhesion of hydrophilic coatings on hydrophobic surfaces, and vice versa, increasing the wettability of the product [116–118]. A widely used natural surfactant in EC is lecithin, which is GRAS and exhibits a high efficiency as an emulsion stabilizing agent [119]. Other common surfactants are lecithin

derivatives, acetylated monoglyceride, ethylene glycol monostearate, glycerol monostearate, and sorbitan fatty acid esters (Tweens), which are all GRAS [120].

### Antimicrobial, Antioxidant, and Anti-browning Additives

ECs can be loaded with different bioactive compounds of natural origin to promote specific functionalities, such as antimicrobial, antioxidant, and anti-browning activities, by retarding deterioration, rancidity, or discoloration due to oxidation caused by free radicals and reducing the extent of enzymatic and nonenzymatic oxidation of phenolic compounds [121–125]. The organoleptic properties of the coated products can also be improved if the coating is loaded with flavoring or coloring agents, as well as with sweeteners, spices, and seasonings [82, 111, 126].

**Essential Oils** Essential oils (EOs), which consist of a mixture of esters, aldehydes, ketones, terpenes, and phenolic compounds, found in aerial or underground parts of plants [127], are gathering an increasing interest because, in addition to their well-documented health-beneficial properties [128, 129], they exhibit a strong nonspecific antimicrobial action [130]. Because of their low solubility in water, EOs need to be delivered through suitable carriers, such as nanoemulsions or other colloidal particles [127, 131]. In the case of ECs, the biopolymer solution might contain macromolecules with an interfacial activity, which is sufficient to stabilize a fine dispersion of EOs; however, in most cases, an additional emulsifier and an adequate emulsification process are required to obtain a coating with homogeneously dispersed EOs [132–134]. However, some authors have reported that the incorporation of essential oils in EC might impart undesirable sensorial modifications in foods [128, 135].

**Plant Extracts** Different plant extracts may exhibit a significant antimicrobial activity, associated to their high content in polyphenols [136], as well as antioxidant and antimutagenic properties, and to inhibit lipid oxidation in food [137]. For example, *Aloe vera* gel, because of its high content in antiseptic compounds, as well as in polysaccharides, has attracted increasing interest in the preparation of active coatings [138–141]. Several other plant extracts have been tested as additives to coating formulations for their antimicrobial activity, such as Tartary buckwheat extracts [142] or olive leaf extracts [143]. A sustainable source of antimicrobial bioactive compounds is represented by agri-food residues, which are increasingly exploited, such as the ethanolic extracts of the leaves and pods of cocoa, the leaves and hulls of coffee [144], or the grape seed extracts [39]. Similar to EOs, the addition of natural extracts to EC may impart a bitter taste and astringent or off-flavor, impairing the acceptability of the product [135, 145].

**Extracts from Algae and Mushrooms** Algae and mushrooms are naturally rich in bioactive compounds, with significant antimicrobial, antioxidant, antiviral, and anti-inflammatory activities [137], such as proteins, antioxidant molecules (e.g., polyphenols, flavonoids, and carotenoids), polyunsaturated fatty acids, and polysaccharides [146]. For example, for the preparation of ECs for strawberries, both *Palmaria palmata* seaweed extracts, in combination with chitosan [147], and fucoidan recovered from *Laminaria japonica* alga [148] were tested.

**Animal-Derived Compounds** Another important class of natural antimicrobial compounds is represented by those derived from animal sources. Chitosan is a polycation biopolymer, naturally present in the exoskeletons of arthropods and crustaceans [149], which has attracted significant interest because of its biodegradability, biocompatibility, bioadhesion, and nontoxicity [150]. Because of its film-forming properties and intrinsic antimicrobial activity, it is highly effective in extending the shelf life of food products, retarding oxygen, moisture, solute transports, and aromas [151]. It must be remarked that chitosan can be obtained also from fungal sources [152] and chitin through diatom photobioreactors [153, 154].

Lysozyme is an enzyme naturally found in mammalian milk and poultry eggs [155], which is usually considered safe to be added directly to food [137]. Its combination with compounds capable of destabilizing the outer membrane of Gram-negative bacteria, such as nisin, EDTA, or EOs, significantly increases the spectrum of its antimicrobial activity [137]. Lactoferrin is a whey glycoprotein, which can bind one or two ferric ions ( $\text{Fe}^{2+}$  or  $\text{Fe}^{3+}$ ), whose antimicrobial activity is reported to cause outer membrane damages to microbial cells [156]. Lactoperoxidase, found in high concentrations in bovine milk [157], exhibits antimicrobial activity mainly against Gram-negative bacteria and is frequently used in combination with chitosan in ECs [158].

**Microbial-Derived Compounds** A variety of compounds of microbial derivation exhibit a significant antimicrobial activity. Nisin is a bacteriocin (antimicrobial peptide) produced by lactic acid bacteria (*Lactococcus lactis* subsp. *lactis*) [137], which is a GRAS food preservative, approved for commercial products [159], because of its antimicrobial activity against Gram-positive bacteria [160]. Lacticin is a two-peptide bacteriocin, also produced by *L. lactis* subsp. *lactis*, but with a stronger activity and higher target specificity against Gram-positive bacteria than nisin [161]. Pediocins, produced by *Pediococcus* species, are small, cationic proteins with antilisterial activity and antimicrobial effect against Gram-positive bacteria, which are reported to maintain their antimicrobial activity over a wide range of pH values and

temperature range [162]. Reuterin is a D-ribose analog generated by *Lactobacillus reuteri* during glycerol metabolism. It has a recognized antibacterial activity against food-borne pathogens and spoilage bacteria, with higher activity against Gram-negative bacteria than Gram-positive bacteria [163].

Organic acids, such as acetic, ascorbic, citric, and lactic acid, are organic compounds, all characterized by one or more carboxyl groups ( $-\text{COOH}$ ) in their structure, and in addition to being GRAS, they are known to possess antimicrobial properties of interest for food application [137].

## Applications of Edible Coatings to Food Products

A survey of the recent scientific literature on the application of ECs for the preservation of different products, summarized in Table 2, includes their effect on shelf-life extension. In general, product applications have focused on the treatment of products characterized by high added value, short shelf life, or difficult logistics associated with production in remote or rural areas. As a recent trend, it must be remarked that natural coating materials are increasingly attracting the interest of researchers, as shown by the wide use of *Aloe vera* gel [138–141] and other natural polysaccharides, such as chitosan [9, 12, 25, 39, 74, 142, 147, 152, 164–166, 168–170, 172, 174, 179, 180, 186, 190, 193, 201], or seaweed extracts [148], which exhibit an intrinsic antimicrobial activity, significantly contributing to extending the product shelf life. The tendency toward natural products is also confirmed by the wide use of natural gums [47, 181], starches [20, 188, 195], cellulose, and its derivatives [32, 167, 173, 175, 194], and other compounds recovered from agri-food residues [144, 183, 203]. Glycerol is frequently used as a plasticizer, at concentrations ranging from less than 1 to 20%. Another trend that became evident in recent years is the addition of natural extracts as antimicrobial or antioxidant agents, including essential oils [9, 12, 29, 172, 177–179, 181, 185, 186, 188, 192, 193, 195, 198, 202] and different plant and fruit extracts [39, 48, 142, 144, 164, 167, 174, 182, 184, 187, 194, 201].

## Pulsed Light

### Fundamentals

PL, also known as pulsed UV light, high-intensity broad-spectrum pulsed light, pulsed white light, or intense light pulses, is a nonthermal technology that has gained increasing interest in the last decade, from both the research world and food processing industry, as an alternative to traditional chemical and thermal microbial decontamination methods of foods, food contact surfaces, and equipment [204, 205].

PL is claimed as an improved version of the UV rays technology, especially for applications requiring rapid and

**Table 2** Edible coatings used for the preservation of different food products, with details on coating formulation

Product	Coating solution	Plasticizer/surfactant	Active ingredient	References
<b>Fruits</b>				
Apricot	<i>Aloe vera</i> gel or basil seed mucilage	–	–	[140]
Blueberries	Gum arabic	Glycerol	Baobab fruit extracts	[48]
Blueberries	Fungal chitosan	Glycerol + Tween 20	Procyanidins from grape seeds	[164]
Dates	Gelatin, chitosan, or guar gum	Glycerol	Olive oil + ascorbic acid	[165]
Figs	Sodium alginate + calcium chloride + chitosan	Lecithin + glycerol	Olive oil	[166]
Grapes	Xanthan gum + calcium chloride	Glycerol + Tween 20	Ascorbic acid	[51]
Guava	Chitosan	–	ZnO nanoparticles	[25]
Guava	Carboxymethyl cellulose + stearic acid	Lecithin	Date seed oil	[167]
Kinnow mandarin	Carboxymethyl cellulose	–	–	[32]
Kiwi	Chitosan hydrochloride	–	–	[168]
Mango	Gum arabic	–	–	[47]
Orange	<i>Aloe vera</i> gel	Glycerol	–	[138]
Papaya	<i>Aloe vera</i> gel + agar gel	–	Ascorbic acid + citric acid	[141]
Peaches	Chitosan	–	Chlorogenic acid	[169]
Pears	Cassava starch reinforced by starch nanocrystals	Glycerol	–	[20]
Strawberries	<i>Aloe arborescens</i> gel	–	–	[139]
Strawberries	Chitosan	–	<i>Palmaria palmata</i> seaweed extracts	[147]
Strawberries	Fucoidan from <i>Laminaria japonica</i> seaweed	–	–	[148]
Strawberries	Chitosan	Glycerol	–	[170]
Strawberries	Maltodextrins + whey protein isolates	PGPR	Water in olive oil emulsion	[171]
Sweet cherries	Sodium alginate + calcium chloride or chitosan	Glycerol	Olive leaf extracts	[143]
<b>Vegetables</b>				
Broccoli florets	Modified chitosan	–	Mandarin oil nanoemulsion	[172]
Brussels sprouts	Sodium carboxymethyl cellulose	Candelilla wax + Tween 40	–	[173]
Green beans	Modified chitosan	–	Mandarin oil nanoemulsion	[9, 12]
Romaine lettuce	Chitosan hydrochloride	–	–	[168]
Rucola	Modified chitosan	–	Lemon oil nanoemulsion	[174]
Spinach	Nanocellulose (from hemp trunks)	–	–	[175]
Tomatoes	Pectin + corn flour	Glycerol	–	[176]
Tomatoes	Ethanol extracts of the leaves and pods of cocoa, leaves, and hulls of coffee	Glycerol + sodium hypochlorite	–	[144]
<b>Fresh-cut fruits and vegetables</b>				
Apple	Sodium alginate + calcium chloride	Glycerol	Thyme EO + ascorbic acid + citric acid	[177]
Cantaloupe melon	Sodium alginate + calcium chloride	Glycerol	Thyme EO + ascorbic acid + citric acid	[178]
Cantaloupe melon	Carboxymethyl cellulose or chitosan	–	Citral nanoemulsion	[179]
Cantaloupe melon	Fungal chitosan	–	–	[152]
Cucumber	Chitosan	–	–	[180]
Eggplants	Soy protein isolate	Glycerol + beeswax	–	[181]
Globo artichoke	Locust bean gum	Glycerol	<i>Foeniculum vulgare</i> EO	[181]
Lotus root	Xanthan gum + chitosan	–	–	[182]
Mango	Sesame proteins + guar gum + calcium chloride	Glycerol	Fresh mango puree	[183]
Papaya	Sodium alginate	Polyoxyethylene sorbitan monooleate	Thyme EO or oregano EO	[29]
Pineapple	Sodium alginate	Glycerol	Citral nanoemulsion	[184]
<b>Fish</b>				
Hake medallions	Whey protein isolates	Glycerol	Oregano EO	[185]

**Table 2** (continued)

Product	Coating solution	Plasticizer/surfactant	Active ingredient	References
<i>Huso huso</i> fillets	Chitosan	–	Fennel EO	[186]
Tilapia fillets	Chitosan	–	Tartary buckwheat extracts	[142]
Trout fillets	Sodium alginate	–	Resveratrol	[187]
Trout fillets	Quinoa starch	Glycerol	Lemon or sage EO	[188]
<b>Meat</b>				
Beef	Gelatin + chitosan	Glycerol	-	[189]
Beef	Chitosan	–	–	[190]
Beef	Whey protein nanofibrils	Glycerol	Titanium dioxide nanotubes	[191]
Chicken breast	Whey protein isolates	Glycerol	Oregano EO	[192]
Chicken breast	Chitosan	Glycerol + Tween 80	<i>Zataria multiflora</i> Boiss. EO	[193]
Pork	Chitosan + gelatin	–	Grape seed extracts	[39]
Pork	Sodium alginate + carboxymethyl cellulose	Glycerol	Epigallocatechin gallate	[194]
<b>Miscellaneous</b>				
Eggs	Sweet potato starch	Glycerol + Tween 80	Thyme EO	[195]
Eggs	Whey protein isolates + pectin + transglutaminase	Sorbitol	–	[196]
Fiordilatte cheese	Sodium alginate + calcium chloride	–	Freeze-dried <i>Lactobacillus rhamnosus</i> + fructooligosaccharides	[197]
Gouda cheese	Zein	–	Laurel EO	[198]
Gouda cheese	Whey protein	Glycerol	Lactoperoxidase + <i>Bunium persicum</i> EO	[199]
Kashar cheese	Whey protein concentrate	Glycerol	<i>Williopsis saturnus</i> killer yeast	[200]
Manchego cheese	Chitosan	Glycerol + Tween 20	<i>Santolina chamaecyparissus</i> L. extracts	[201]
<i>Pholiota nameko</i> mushroom	Sodium alginate	Glycerol	Thyme EO + L-cysteine + nisin	[202]
Shiitake mushrooms	Chitosan + guar gum	Glycerol	–	[74]
Goat and Tybo cheese	Agar	Glycerol	Enterocins from <i>E. avium</i>	[18]
Walnut kernels	Walnut flour protein	Glycerol	–	[203]

effective disinfection treatments. UV radiation is the part of the electromagnetic spectrum of sunlight, in the range 100–400 nm, which can be subdivided into UV-A (315–400 nm), UV-B (280–315 nm), UV-C (200–280 nm), and the vacuum UV range (100–200 nm). UV-C has the strongest germicidal effect and is widely used to inactivate microorganisms [206]. The classical UV-C treatment works in a continuous mode, called continuous wave (CW)-UV light, and typically involves the exposure (from minutes to hours) of the substrate to be disinfected to low power (5–80 W) monochromatic light (254 nm) emitted by a low-pressure mercury lamp [204]. In the PL process, instead, food or nonfood materials are exposed to a successive repetition of short (100 ns to 1 ms) high-intensity pulses (flashes) of polychromatic light (180–1100 nm), including UV (180–400 nm), visible (400–700 nm), and infrared (700–1100 nm) regions, emitted by an inert gas (e.g., xenon) lamp [13]. The light used for food processing applications is typically pulsed at 1 to 20 flashes per second at an energy density in the range of about 0.01 to 50 J/cm<sup>2</sup> at the surface [13].

The inactivation effectiveness of PL has been tested against a great variety of pathogenic and spoilage microorganisms, including bacterial species (both as vegetative cells or spores), yeast, fungi, and viruses spread on agar, food, or food contact surfaces or suspended in fluids (air, water, liquid foods) [13, 207–209].

The lethal action of the PL process is attributed to the effects of the high peak power and the UV component of the broad spectrum of the light flashes, which result in the coexistence of different inactivation mechanisms. In particular, the UV component of PL can be absorbed by DNA and other components of the cell, thereby causing photochemical damage, which kills microorganisms [210]. Moreover, PL spectrum includes visible and near-infrared regions, which convey heat to the surface of the processed substrate, inducing a local instantaneous increase of the temperature of a thin (a few  $\mu\text{m}$  thick) surface layer only to an extent sufficiently high to destroy microbial cells (photothermal effect) [211, 212]. Additionally, structural damages caused by the high-power pulsing effect (photophysical effect) have been also detected through microscopy detection as well as the quantification of

the leakage of intracellular matters, such as proteins, which was not observed when the same microbial cells were exposed to CW-UV light treatment [213, 214]. The relative importance of each mechanism may depend on the peak power of the light pulses, the composition of the emitted light spectrum, and the type of microorganism as well as the optical properties of the target substrate, among others.

The occurrence of this multitarget microbial inactivation mechanism, along with the high emission power, which is likely to increase the capability of PL to penetrate the treated substrate, can explain the generally reported higher sterilization effectiveness of PL in comparison with CW-UV [214]. Moreover, PL is a fast and easily operating, residue-free decontamination technology, which is characterized by low energy consumption, can be combined with other disinfection methods, and is suitable for integration in industrial processing lines, enabling high product throughput.

## Technical Aspects and Pulsed Light Equipment

### Definitions and Terminology

Before providing information concerning PL equipment and processing, it is worth giving the definition and units of the main parameters characterizing this technology [215].

- Pulse duration (or pulse width) is the time interval (ns to ms) during which the light energy is delivered.
- Number of pulses (or flashes) is the total number of pulses or flashes of light delivered to the target substrate.
- Pulse repetition frequency is the number of pulses of light delivered per second (Hertz [Hz]), commonly expressed as pps (pulses per second). Due to the design features of the lamp, in PL systems, the pulse frequency is typically limited to a few Hertz. However, higher pulse frequency can be achieved by using two or more lamps, placed and flashed in sequence.
- Exposure time is the actual time (in seconds) of exposure of the substrate to the light flashes and is calculated as the number of pulses times the pulse width.
- Peak power,  $\phi$ , is pulse energy divided by the pulse duration and (in watts (W)).
- Fluence rate ( $F_{\circ}$ ) is the radiant power passing from all directions through an infinitesimally small sphere of cross-sectional area,  $dA$ , divided by  $dA$  (in  $W/m^2$ ).
- Fluence ( $F$ ) (or PL dose) is the total radiant energy from all directions passing through an infinitesimally small sphere of cross-sectional area  $dA$ , divided by  $dA$ , for a certain time (in  $J/m^2$ , even though in PL technology, it is often expressed in  $J/cm^2$ ). The fluence is the fluence rate multiplied by the exposure time. However, when the substrate exposed to PL treatment is a given volume of a liquid containing a certain microbial load expressing the

PL dose as total radiant energy per unit area is not correct. Thus, several authors have proposed that the fluence (in  $J/L$  or  $J/mL$ ) is correctly defined, according to Eq. 1, as the PL energy output (radiant power,  $\Phi$ , in W) delivered to a volumetric flow ( $Q$ , in L/s, or mL/s):

$$F = \frac{\phi}{Q} \quad (1)$$

The evaluation of the operational costs based on the volumetric electrical energy input ( $J/L$  or  $J/mL$ ) allows comparing the results of PL treatments carried out with different PL units as well as with those obtained utilizing other inactivation technologies [206].

### Pulsed Light Generation and Equipment

PL is generated using pulsed power technologies that involve the generation of high-power electrical pulses and their transformation into high-power light pulses. The systems used for PL applications typically include a power/control module, a lamp housing with a flash lamp, a treatment chamber, and an auxiliary equipment, such as pumps, cooling systems, and devices to measure temperatures and fluence rate or fluence.

The power/control module is used to manage the process, start the flashes, control the treatment time, and modulate the generation of high-power electrical pulses to obtain the desired configuration of the pulse energy and rate. In particular, during PL treatment, the power/control module converts alternating current (AC) into direct current (DC), cyclically accumulated in a capacitor bank, and released by a special switch as high-voltage electric pulses through the lamp unit. The latter consists of one or more lamps with quartz envelopes, each equipped with two electrodes, typically filled with inert gases (e.g., xenon, krypton). The inert gases effectively convert the pulsed electrical energy into a broad spectrum of pulsed radiant energy. Approximately 25% of the wavelength lies in the UV range, 45% in the visible range, and 30% in the infrared range. Moreover, the shape (linear, spiral, etc.) and the size of the lamp can be customized for the specific application to ensure uniform irradiation of the target surface (<http://www.xenoncorp.com/>).

The PL equipment can be operated either in batch or in continuous flow mode, as schematized in Fig. 1.

Batch systems are the most widespread and used for preliminary investigations at the laboratory scale on the effects of the main PL treatment parameters on the decontamination of liquid and solid products. Figure 1a shows a schematic of a typical laboratory-scale batch system. It consists of a chamber in which the solid or liquid sample is placed on an adjustable tray that allows regulating the distance between the sample and light source. A housing lamp with a xenon lamp is generally mounted on the upper part of the treatment chamber.



and the input. With increasing the intensity of these parameters or decreasing the distance of the product from the lamp, the total fluence increases and, hence, microbial inactivation is enhanced [204, 208, 222, 223].

Appropriate PL dosimetry is fundamental to compare results and for process scale-up [215]. In this regard, fluence measurements should be standardized and precaution should be taken in reporting the energy dose at the substrate surface or within the substrate, which is substantially different from the energy delivered by the light source. This is because of the radiation loss attenuation across the path length of the treatment medium due to light absorption and scattering phenomena [221, 224].

The composition of the PL spectrum is another important process parameter determining the antimicrobial action of PL technology [217, 225–227]. Although the effectiveness of the full spectrum of high-intensity PL has been widely demonstrated to be effective against different microbial species, the photochemical effect of UV-C plays a major role in microbial inactivation. Therefore, depending on the specific application, the use of wavelength filters, such as solid filters (e.g., glass UV filters) or liquid filters (e.g.,  $\text{CuSO}_4$  solution) [209, 217], or the adjustment of electrical current [228] allows selecting the most suitable wavelength ranges emitted by the lamp to achieve the desired microbial inactivation while avoiding or minimizing alterations of the substrate properties.

The temperature increase in the substrate exposed to PL treatment is mainly a function of the total amount of energy delivered by the light source to the target, as well as of the pulse frequency and composition of the PL spectrum. Preventing overheating is of utmost importance to avoid seriously compromising food quality, especially during long processing time. To this purpose, an efficient cooling system is incorporated in the equipment, and an appropriate pulse repetition rate and distance between the product and the lamp source must be selected [208].

**Design Parameters** Geometry and setup of the treatment chamber, lamp, and deflector as well as the number of lamps significantly influence the treatment uniformity and, hence, the efficiency of the PL process and product quality.

**Product Properties** The composition and physical properties of the substrate exposed to PL treatment, which can enhance absorption, reflection, or scattering phenomena of the incident light or induce shading effect, affect the effective radiation dose reaching the target, thus biasing treatment effectiveness and uniformity. In this regard, proteins and fat-rich substrates are unsuitable for PL treatment, since these components can competitively absorb light thus decreasing PL inactivation effectiveness, while light absorption does not occur in carbohydrates [222]. Therefore, food products with high protein and fat content have little potential to be efficiently

decontaminated by PL, while vegetables and fruit are eligible for PL treatment [204].

Different physical properties of the substrate are playing a role when flashing solid or liquid foods. Opacity, turbidity, coloring compounds, viscosity, and suspended particles may significantly hinder either the penetration of light pulses and light absorption into the liquid food causing a significant decrease in PL treatment effectiveness and uniformity [208, 221]. Therefore, while PL can be successfully applied to transparent liquid (e.g., drinking water and clear fruit juices), in opaque liquid foods, such as orange juice or milk, the effect of PL might be limited only to the superficial layer of the substrate. Moreover, solid product properties, including topography, reflectivity, hydrophobicity of the treated surface, color, and opacity, greatly influence the successful PL microbial inactivation [209, 229]. For example, surface roughness, crevices, or pores may shade or hide microbial cells during treatment. The hydrophobicity of the food surface may affect the distribution of microbial cells promoting the formation of cell clusters and reducing PL inactivation. Furthermore, a high surface reflectivity, causing a decreased light absorption of the microbial cells, could lead to poor inactivation [229, 230].

The limitations to the use of PL technology in the decontamination of solid matrices are common to other methods applied in the food industry to decontaminate raw fruits and vegetables or packaging material, such as washing with solutions containing chlorine, peracetic acid, or hydrogen peroxide [205, 209]. The challenge is, therefore, represented by engineering solutions as the re-design of the equipment to promote more uniform surface irradiation of rough surfaces of solid and opaque liquid products [13, 204]. Additionally, absorption-enhancing agents (e.g., carotenoids, fat), sprayed on the surface or added to the formulation of foods, have been suggested to maximize the absorption of the bactericidal wavelength of PL [209, 231].

**Microbial Factors** The lethal effect of PL also depends on the intrinsic properties of the microbial cells, namely the type of microorganism, growth phase, and inoculum size. For example, it is widely recognized that PL susceptibility of microorganisms exhibits the following trend: Gram-negative bacteria < Gram-positive bacteria < yeasts < bacterial spores < molds < viruses [13]. The growth phase is another microbial factor which may affect the light sensitivity of microorganisms. In general, microbial cells in stationary growth phase show greater resistance to PL than cells in either lag or exponential phase [222]. Moreover, PL decontamination is less effective in highly contaminated products due to light attenuation phenomena [204]. In such a case, microorganisms may overlap or aggregate, the outer microorganisms shading those located underneath or inside the cluster [222, 232], and, consequently, non-uniform treatments and reduced PL inactivation effectiveness are likely to occur.

In conclusion, there is a need for more systematic and accurate studies on the effect of the most relevant factors affecting the successful application of the PL process, which should include also the standardization of the measurements of the treatment dose as well as of the experimental procedures used by different research groups. Additionally, the design of PL systems should be optimized for each application (e.g., inducing a turbulent flow or enhancing the mixing conditions) to improve the treatment homogeneity and, thus, increasing the microbicidal effects of this technology. Finally, the need of integrating proper cooling systems in the treatment chamber should be emphasized to minimize temperature buildup during the pulse treatment.

### Main Applications to Food Products

PL treatment is a fast, environmentally friendly, nonthermal technology with many potential applications in the food industry for food processing and food contact surface decontamination. In this regard, the technology appears especially interesting for the decontamination of food packaging material, fresh and fresh-cut produce prior to or after packaging, as well as for the stimulation of fruit physiology to promote the production of functional compounds. Since 1996, PL irradiation has been approved by the US Food and Drug Administration to decontaminate food or food contact surfaces, provided that the treatment uses a xenon lamp with an emission wavelength between 200 and 1000 nm, a pulse duration not exceeding 2 ms, and a cumulative energy level not exceeding 12 J/cm<sup>2</sup> (Code of Federal Regulation, CFR: 21CFR179.41 [233]).

Despite the recent advancements of PL technology, the increasing number of PL manufacturers, and the number of studies on surface decontamination, currently, PL has been successfully applied at the industrial scale only to decontaminate food packaging materials (<http://www.claranor.com/>). The utilization of PL treatments at the pilot or industrial scale to extend shelf life and improve the quality of food products is still lacking.

The capability of PL to inactivate microorganisms, either artificially inoculated or naturally present on the surface of food products, before packaging has been extensively investigated on a large number of fruits and vegetables. The results reported in the literature demonstrated that, although complete inactivation is not possible, PL treatments allowed obtaining from 1 to 6 log reduction of the microbial load while preserving quality attributes of foods [13]. The different inactivation levels reported often arise from equipment type and configuration, experimental protocols, and fluence measurements other than individual characteristics of the microbial strains.

For example, Bialka and Demirci [234], by applying a PL treatment at fluences of 23–59 J/m<sup>2</sup>, observed a reduction of 2–4 log CFU/mL of *Escherichia coli* O157:H7 inoculated on the surface of blueberries, raspberries, and strawberries. In

another study, Aguiló-Aguayo et al. [235] assessed the impact of PL, at a fluence of 2.3 or 5.4 J/cm<sup>2</sup>, on surface decontamination of native microflora or inoculated *Saccharomyces cerevisiae* of red-ripe tomatoes. PL treatment was more effective against the inoculated microorganisms (2.3 log CFU/mL reduction) than total microflora (1 log CFU/mL reduction). The authors also investigated the impact of PL exposure on physicochemical (color, texture, weight) and nutritional properties of tomatoes during storage at 20 °C for 15 days. Interestingly, they found that PL treatment did not induce any changes in physicochemical properties of tomatoes, while the nutritional properties of the PL (30 J/cm<sup>2</sup>)-treated samples remained unaffected (ascorbic acid) or were improved (total lycopene,  $\alpha$ -carotene, and  $\beta$ -carotene).

The *in-package* decontamination of food products with PL is also of great interest for the future commercial application of this technology, since it allows the treatment of food already packed, avoiding undesired post-treatment recontamination. Nevertheless, only a few publications, so far, focused on *in-package* decontamination of food products, most of them referred to decontamination of meat and meat products, fish, and, to less extent, of fruits and vegetables [236]. Therefore, more research efforts are required to assess the feasibility of *in-package* PL processing of foods. Future studies should investigate not only the effects of physical properties of the products, processing conditions, and susceptibility of the native microflora to PL exposure, but also the chemical (e.g., composition), physical (e.g., thickness), and optical properties (e.g., UV transmissivity) of the packaging films. Moreover, particular attention should be paid to evaluate the structural and barrier property changes of packaging and migration of compounds from packing materials to foods induced by PL.

Finally, several recent studies reported the potential of PL in modulating the metabolic activity of fresh produce [237, 238]. The main findings were the delay of senescence and deterioration of fruits and vegetables during storage in MAP and, interestingly, the stimulation of plant natural defenses against fungal diseases and the biosynthesis of bioactive antioxidant compounds [216, 237–239]. For example, it has been shown that PL exposure can significantly increase lycopene content in tomatoes [216], the amount of anthocyanins and phenolic compounds in figs [240], the phenolic compounds in apples [239] and persimmons [241], and the vitamin D<sub>2</sub> in mushrooms [242]. It is particularly noteworthy that the industrial implementation of this kind of PL treatment already took place [243].

### Combined Edible Coating and Pulsed Light Treatments

Several methods have been proposed for the preservation of fresh products, which could allow better retention of their

quality attributes and overcome the limitations and drawbacks of traditional thermal treatments, especially their strong impact on the nutritional, functional, and sensorial properties of products [244]. Some of the nonthermal preservation methods that have been investigated include the deposition of ECs loaded with additives (active EC), the use of controlled atmosphere storage or modified atmosphere packaging, high pressure, PL, pulsed electric fields and ultrasounds treatments, and ionizing radiations [245]. However, often the hurdle approach is more useful and successful than the single method, the two main advantages relying upon the superimposition of the effects of the different preservation factors and to the possibility to tune the more expensive stress to lower intensities, which positively affect the costs and the energy expenditure of the innovative technology [11].

In the “[Edible Coatings](#)” and “[Pulsed Light](#)” sections, the advantages of using active ECs and PL for food preservation have been outlined, including the reduction of the incidence of food-borne pathogens and spoilage microorganism, either through their inactivation or growth inhibition, the extension of the shelf life of the produce, and the prevention of food quality losses along the distribution chain. By combining these two hurdles, efficient food preservation methods can be developed, exploiting the different antimicrobial mechanisms of the two technologies and different time scales of operation (PL causes an immediate microbial reduction on the food surface, while active ECs ensure the inhibition of microbial growth over an extended time).

ECs and PL treatment can be applied to the product in two different sequences: EC followed by PL treatment of foods (before packaging or eventually already *in-package*), or PL treatment applied to the product which is then covered with the EC. Following the first approach, residual or cross-contamination from the coating surface is avoided, contributing to extend the shelf life of the product. The application of PL on coated products facilitates their manipulation allowing a better exposure to light pulses of the entire surface, which is not possible for packaged products. Despite the screening effects of the packaging could reduce PL treatment effectiveness, the exposure of foods to light flashes after packaging is particularly advantageous to extend the shelf life of products due to the removal of residual contaminations both on foods and package.

Following the second approach, PL is used to decontaminate the surface of the products before their stabilization obtained by applying the EC. This strategy is particularly useful when a significant reduction of the microbial population on the product surface is needed. In this case, cross-contamination should be avoided, and therefore, edible coatings must be applied through spraying or electro-spraying techniques. Moreover, following this approach, the process becomes more laborious, being two stages of drying or dripping off from the product

necessary before the PL treatment and after the coating application.

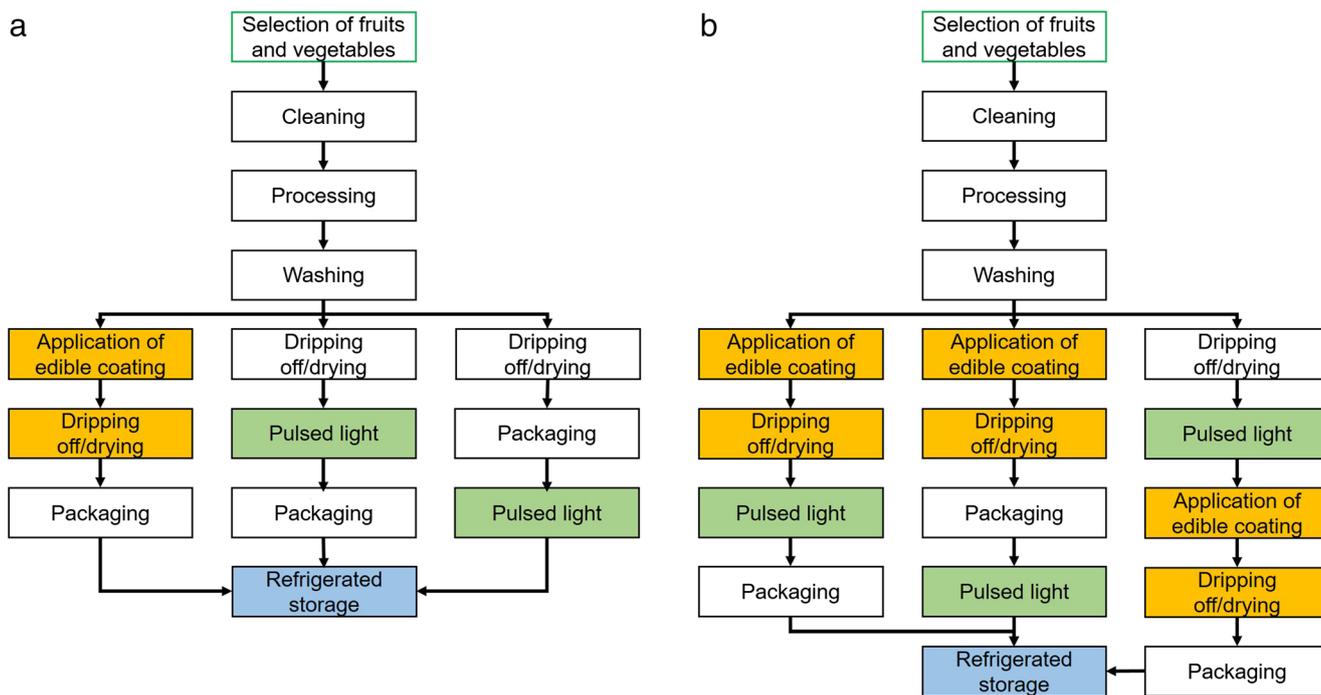
These two alternative processing strategies are depicted in Fig. 2, reporting a generic processing line for fruit and vegetable transformation [246] in which the EC and PL preservation methods are implemented individually (Fig. 2a) or in combination (Fig. 2b). Fresh-cut fruits and vegetables are very critical to be processed. The cutting surface, consisting of wounded tissues, is extremely susceptible to softening, browning, and discoloration phenomena [247] and, due to the release of the intracellular compounds, is exposed to microbial and enzymatic attack [174]. However, fresh-cut fruits and vegetables are a value-added convenience food in high demand for consumers oriented toward natural and healthy products. Therefore, the use of the hurdle approach, as shown in Fig. 2b, to extend their shelf life is fully justified and understandable, and most of the studies on the use of combined PL and ECs have been targeted on fresh-cut fruit and vegetable preservation, as discussed in the “[Transmittance of PL Through the Coating Layer](#)” section.

## Possible Interactions Between PL Treatment and Coating Application

### Effect of PL on Coating Properties

A first important aspect to consider in the combination of PL with the EC is concerned with the possible modifications of the coating when exposed to a PL treatment.

Previous studies have reported that the main modifications observed in edible films subjected to PL can be related to the occurrence of cross-linking reactions, due to photo-polymerization effects, and film deterioration phenomena, such as depolymerization and retrogradation, whose relative extent depends strongly on film composition. For example, taro starch-based films, characterized by a low amylose content, when exposed to PL, underwent mainly a physical modification through photo-polymerization, with the induced slight cross-linking that improved starch–glycerol interactions [248]. Conversely, cassava starch-based films, characterized by high amylose content, did not show any evidence of cross-linking but exhibited a measurable deterioration due to a photodegradation effect [248]. Other authors reported that PL-treated starch-based films exhibited an increase in crystallinity, reduced chain mobility, and formed more closed structures, independently of their amylose content [249, 250]. Furthermore, techno-functional characterization of PL-treated starch-based films showed that, under the conditions of maximum fluence permitted by the US FDA (12 J/cm<sup>2</sup>), photodegradation reactions prevailed, independently on amylose content of starch, and PL-treated films exhibited a significant deterioration of their properties, as shown by an increase in contact angle, surface roughness, and crystallinity and a



**Fig. 2** Process flow diagram of fresh-cut fruit and vegetable processing with the inclusion of edible coating application or pulsed light treatments, individually (a) and combined (b)

decrease in tensile strength, transparency, and water content [251]. The degradative effect of PL treatment was observed also for casein-based films, which exhibited a significant deterioration of mechanical properties, e.g., a reduction in Young's modulus and maximum stress values, and surface characteristics, e.g., increase in wrinkle density and film opacity [252]. In contrast, no measurable effects on mechanical properties, such as tensile strength, elongation at break, and Young modulus, were recorded for starch-based films used as packaging for cheddar cheese, upon application of a PL treatment at a fluence varying in the range from 1 to 12 J/cm<sup>2</sup> [253].

These partially contradicting data derive from the limited number of studies on the effect of PL treatment on edible films and coatings, which prevents to draw general conclusions. The analysis of the data reported for PL treatment of packaged products can, however, provide some additional elements. When a packaging material is subjected to PL, it is important to monitor not only the physical and mechanical stability of the films but also the potential chemical migration, which might cause safety concerns [254]. Changes in mechanical properties, such as reduced strength or extensibility, and impact strength or cracking are generally reported, and the alteration of the mass transfer properties, namely permeation, migration, and scalping, can take place [255]. However, these phenomena are likely to occur upon intense PL treatments, whereas *in-package* treatment carried out at milder conditions does not cause any measurable effect on film properties [254]. For example, polypropylene films, treated with PL at 1.27

J/cm<sup>2</sup> fluence, showed no significant changes in elastic modulus, yield strength, percent elongation at yield point, maximum tensile strength, and percent elongation at break [256].

In conclusion, due to the lack of available data, no prediction can be made on the effect of PL on the techno-functional properties of ECs, and preliminary experiments should be carried out to verify their compatibility. In any case, the risks associated with the migration of the coating components into the foods are less critical than for polymeric films, where, instead, potentially toxic compounds are present.

#### Transmittance of PL Through the Coating Layer

A second important aspect to consider is the transmittance of PL through the coating layer. No specific studies are available on ECs, and therefore, some considerations can again be made on the basis of the literature data concerning the PL treatment of packaged products. As a general recommendation, a successful *in-package* application of PL requires the selection of proper packaging materials, characterized by high light transmissibility, in particular of the UV fraction. Therefore, polymeric matrices that may interfere with light absorption because of their intrinsic nature or because they contain additives, such as polyphenols or anthocyanins, should be avoided [254].

As already mentioned, in PL treatments, UV light below 270 nm is of greatest relevance for the decontamination processes [226]. Therefore, materials with a cutoff wavelength, which is defined as the wavelength below which the light

transmission is negligible (absorbance of 1.0), below 180 nm, such as polyethylene (PE) and polypropylene (PP), are very suitable for *in-package* applications, followed by polyvinyl chloride (PVC) and polyamide (PA) with a cutoff of about 240 nm, polystyrene (PS) and polycarbonate (PC) with a cutoff comprised between 270 and 280 nm, and polyethylene terephthalate (PET) with a cutoff of 310 nm (all given for 10- $\mu\text{m}$ -thick films [254]). Moreover, a higher degree of polymer crystallinity, the presence of morphological inhomogeneity, and the inclusion of additives are also reported to affect light transmittance [254, 257]. Previous studies have shown that a 12- $\mu\text{m}$  polyethylene film, a 48- $\mu\text{m}$  polyamide/polyethylene/vinyl acetate-based copolymer, and a 60- $\mu\text{m}$  polyamide/polyethylene copolymer did not cause any reduction in the PL treatment (fluences of 0.175 and 0.35  $\text{J}/\text{cm}^2$ ) effectiveness against *Listeria monocytogenes* [258]. Similarly, tests carried out on wrapped products showed that PL was suitable for the treatment of packaged chicken frankfurters in PP films [256] and chicken breast in different films (e.g., PP, PVC, PET/PP) [259, 260].

In the case of edible films, it was demonstrated that PL, with a fluence comprised between 9.2 and 12.3  $\text{J}/\text{cm}^2$ , can pass through starch-based films without losing effectiveness, as light in the range 200–1100 nm is not adsorbed [253]. However, some light absorption was recorded when the films were loaded with antimicrobial compounds, such as sodium benzoate and citric acid [253]. In the case of gellan gum-based EC, enriched with apple fiber and used on fresh-cut apples, the transmittance values of UV-A, UV-B, and UV-C radiation, determined for a film thickness of 155.75  $\mu\text{m}$  [261], were 99.3%, 99.0%, and 73.0%, respectively, showing that the coating blocked a significant part of the incident UV-C radiation [262].

The lack of sufficient data on the transmittance of PL through ECs, especially as a function of layer composition, thickness, and eventual presence of additives, prevents from drawing a general conclusion also in this case. However, if the contribution of PL treatment applied after the deposition of the EC is to remove any eventual contamination that occurred on the film surface, the issue of light transmittance becomes not important. Instead, the protection of the food surface from PL radiation might also be appreciated for some photosensitive products.

### Main Applications to Food Products

Based on the considerations reported in the “[Combined Edible Coating and Pulsed Light Treatments](#)” section, it is comprehensible that, to date, only a limited number of studies have focused on the utilization of PL in combination with the application of ECs to inhibit microbial growth and to extend the shelf life of food products.

Table 3 reports a survey of the published scientific papers on combined treatments, presenting the main findings for different products treated and providing details on the formulation and application of ECs, PL treatment conditions, and on the sequence of the treatments.

Interestingly, most of the available data concern fresh-cut fruits and vegetables, which represent, as already discussed, an especially critical product characterized by a high added value. Only one article treated the preservation of green beans, which were considered more like a model food (for the comparison with other technologies, across different laboratories [8, 9, 12]) than for the industrial interest for preservation purposes.

In the data of Table 3, all the different strategies discussed in Fig. 2 were investigated, with the prevalence of coating application before PL treatment [8, 262–264, 266, 267] and of coating application followed by packaging and PL treatment on the packaged product [265, 268, 269].

The PL treatment carried out after coating application is generally preferred when the starting material has a low initial microbial load, as in the case of fresh-cut fruits and vegetables, if properly processed, and resulted to provide a significant contribution to extending the product shelf life, through the preservation of quality parameters and the inhibition of microbial growth. In addition, in some cases, it was also reported that there was an increase in the polyphenolic content of food products, as a consequence of the abiotic stress induced by PL [266].

Remarkably, only two papers considered the application of PL before coating application, in comparison with the application of coating before PL treatment [264, 266]. The application of PL before EC ensured a higher initial microbial inactivation, as shown for fresh-cut mango slices, which can be explained by the fact that the microbicidal effect of PL is not screened by the presence of the coating, which both might exert a protective role on microorganisms and reduce the transmittance of radiation. Interestingly, during a 14-day product shelf life, better quality parameters and microbial stability were observed for the product treated with PL before the coating application [264]; however, the coating application before PL treatment ensured a higher total phenolic content [266].

### Conclusions and Perspectives

Consumers' quest for natural products delivering a high level of convenience has promoted the use of mild preservation technologies for ensuring product safety and retaining the quality attributes of fresh products. Within this frame, EC and PL treatments have recently emerged as very promising for food categories such as perishable high value-added products or fresh-cut fruits and vegetables.

**Table 3** Combined pulsed light and edible coating treatments, with details on pulsed light conditions, formulation and application of edible coatings, the order of the treatments, and food storage conditions

Fresh produce	PL treatment conditions	EC		Order of treatments	Storage conditions	Effect of combined treatments	Ref.
		Film-forming solutions	Method of application				
Fresh green beans	Fluence of 3–6–12 J/cm <sup>2</sup> for each side	1% chitosan + 0.05% mandarin EO	Spraying for 5 s on each side at 1.4 × 10 <sup>-5</sup> Pa	1. Coating 2. PL	4 °C in sterile pouches	The combination of PL treatment and bioactive EC did not show any synergistic or additive antimicrobial effect against <i>L. innocua</i> during storage, with browning spots formation on the samples. However, treatment combination did not affect the samples' firmness during storage nor the coating integrity.	[8]
Fresh-cut apple	Fluence of 12 J/cm <sup>2</sup>	0.5% gellan gum + 0.2% apple fiber + 0.6% glycerol (1 <sup>st</sup> solution) 2% calcium chloride (2 <sup>nd</sup> solution)	Cross-linking dipping for 2 min in each solution, with drip off for 1 min between the two	1. Coating 2. PL	4 °C in the dark	The use of coating incorporating apple fiber followed by PL treatment reduced softening and browning and preserved the antioxidant value of fresh-cut apples. However, the use of ECs reduced the extent of surface decontamination by PL.	[262]
Fresh-cut apple	Fluence of 12 J/cm <sup>2</sup>	2% pectin + 0.7% apple fiber + 1.5% glycerol (1 <sup>st</sup> solution) 2% calcium chloride (2 <sup>nd</sup> solution)	Cross-linking dipping for 2 min in each solution, with drip off for 1 min between the two	1. Coating 2. PL	4 °C in the dark	The combination of both technologies led to a significant reduction in the counts of spoilage microorganisms and higher antioxidant activity values, although an additive effect of both treatments could not be observed.	[263]
Fresh-cut mango	Fluence of 8 J/cm <sup>2</sup>	2% sodium alginate (1 <sup>st</sup> solution) 2% calcium chloride (2 <sup>nd</sup> solution) 2% DL-malic acid	Cross-linking dipping for 2 min in each solution  Dipping for 2 min	1. Coating 2. PL and vice versa	4 °C in polypropylene trays	An additive effect on microbial load reduction by combined treatments was observed. Moreover, the order in which such treatments were performed played an important role in fresh-cut mango preservation. PL treatment followed by coatings ensured better quality parameters and microbial stability with minimal quality deterioration throughout storage.	[264]
Fresh-cut cantaloupe melon	Fluence of 11.7 J/cm <sup>2</sup>	1% chitosan + 1% glycerol + 0.025% sunflower 1% pectin/sodium alginate + 1% glycerol + 0.025% sunflower (1 <sup>st</sup> solution) 2% calcium chloride (2 <sup>nd</sup> solution) 0.5% gellan gum + 0.5% glycerol + 0.025% sunflower (1 <sup>st</sup> solution)	Dipping for 2 min  Cross-linking dipping for 2 min in each solution, with drip off for 1 min between the two	1. Coating 2. PL on a packaged product	4 °C in a polypropylene bag	The combination of pectin, sodium alginate, and gellan gum coatings with PL treatment was effective to reduce fluid loss and retain firmness and maintain the desired headspace gas composition throughout storage. Moreover, a combination of PL treatment and alginate was the most effective treatment condition to extend the shelf life in terms of microbiological quality. The sodium alginate coating adhered well to the surfaces of samples and significantly reduced fluid loss and enhanced firmness compared	[265]

**Table 3** (continued)

Fresh produce	PL treatment conditions	EC		Order of treatments	Storage conditions	Effect of combined treatments	Ref.
		Film-forming solutions	Method of application				
		2% calcium chloride (2 <sup>nd</sup> solution)				to samples treated with PL alone while retaining its physicochemical and nutritional quality.	
Fresh-cut mango	Fluence of 8 J/cm <sup>2</sup>	2% sodium alginate (1 <sup>st</sup> solution) 2% calcium chloride (2 <sup>nd</sup> solution) 2% DL-malic acid (3 <sup>rd</sup> solution)	Cross-linking (1 <sup>st</sup> and 2 <sup>nd</sup> solutions) and layer-by-layer coating deposition by dipping for 2 min in each solution	1. Coating 2. PL and vice versa	4 °C in polypropylene trays	Both combined treatments seem to enhance phenolic compounds content while preserving both antioxidant capacity, phenolic compound content, and color. Moreover, the combination of coating followed by PL treatment enhances mangiferin content by inducing a stress response throughout storage days.	[266]
Fresh-cut cucumber	Fluence of 4–8–12 J/cm <sup>2</sup>	2% chitosan + 0.03–0.08% carvacrol EO	Dipping for 3 min	1. Coating 2. PL	25 °C in a safety cabinet	The combination of the two technologies resulted in a strong synergistic effect, with <i>E. coli</i> reduction when more intense PL treatment was combined with the coating suspension containing 0.08% carvacrol.	[267]
Fresh-cut cantaloupe melon	Fluence of 11.7 J/cm <sup>2</sup>	0–2% sodium alginate + 0–2% glycerol + 0.025% sunflower oil (1 <sup>st</sup> solution) 2% calcium chloride (2 <sup>nd</sup> solution)	Cross-linking dipping for 2 min in each solution, with drip off for 1 min between the two	1. Coating 2. PL on a packaged product	4 °C in a cooler box	The combined treatment increases the shelf life in terms of microbiological quality and reduces the fluid loss while maintaining its physicochemical and nutritional quality. Therefore, combined treatment is necessary due to their synergistic effect since PL treatment contributed to the increased microbiological quality while coating targeted on the improvement of physical quality in fresh-cut cantaloupe.	[268]
Fresh-cut cantaloupe melon	Fluence of 11.7 J/cm <sup>2</sup>	1.86% sodium alginate + 1.47% glycerol + 0.025% sunflower oil (1 <sup>st</sup> solution) 2% calcium chloride (2 <sup>nd</sup> solution)	Cross-linking dipping for 2 min in each solution, with drip off for 1 min between the two	1. Coating 2. PL on a packaged product	4 °C in a polypropylene bag	The combination of alginate coating with PL treatment maintains low lactic acid concentration and retains total aroma compound concentration as storage time increased. Overall, the combination of alginate coating and PL treatment was effective to maintain the fresh-like sensory quality of fresh-cut cantaloupes with minimal changes on sugar contents, organic acid contents, and total aroma compound concentration.	[269]

ECs loaded with active compounds protect food products from deteriorative processes, such as oxidation, water loss,

and browning, and slow down or inhibit microbial growth, while minimally affecting product appearance. PL treatments

are fast and extremely effective for surface decontamination and can be used to treat packaged foods. These technologies can be coupled, in a hurdle approach, with the application of PL treatments, either as a preliminary food decontamination process, to reduce the initial microbial load, or after coating application, and even better after food packaging, to inactivate microorganisms contaminating the product after applying the EC. The latter two options are appealing for those products, for which the initial contamination level is not an issue (e.g., fresh-cut fruits and vegetables), and for which the sequence of coating application and subsequent PL treatment might significantly extend the product shelf life.

A fundamental understanding of the possible synergistic or antagonistic effects of EC–PL treatment combinations needs to be better addressed. For example, the effects of PL treatments on the physicochemical properties of the ECs have not been fully elucidated. Positive effects are likely since the UV components of the PL might cause cross-linking reactions in the coating layer, improving its mechanical properties. Also, the interference of ECs, especially if loaded with bioactive compounds, on light transmission through the coating needs to be better understood. If the PL treatment is carried out after coating deposition, an eventual UV shielding effect might be advantageous if the food surface is particularly light-sensitive, for food protection and coating decontamination.

The process complexity and investment costs that the combined use of EC application and PL treatments bring along clearly represent a potential barrier to a wide industrial application. However, the availability on the market of small-scale pulsed light units and the possibility to develop relatively simple batch systems for coating application makes the proposed combined process especially suitable for high value-added products, especially if produced in rural areas, where efficient logistics is not available. The scaling up of the combined process can be also easily predicted, provided that suitable product conveyors, allowing light penetration, and PL systems with flash lamps in series, allowing to supply the right dose and residence time, are designed and set up. Moreover, since it is possible to apply the PL treatments in continuous mode, this makes it feasible to have considerably high product throughputs. To this purpose, some of the techniques for EC application and PL apparatus already in use for PL industrial applications eventually modified ad hoc to process coated fruits and vegetables are already available and could be used for industrial proof-of-concept tests. Once a few industrial applications have been developed, the economic feasibility for larger-scale applications can then be better assessed to make this method for high value-added convenience food preservation a means to satisfy the consumer's demand for milder processes and increased use of natural ingredients while meeting their quality and safety expectations.

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## Compliance with Ethical Standards

**Conflict of Interest** The authors declare that they have no conflict of interest.

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